

NAMING COMPOUND AND WRITING FORMULAS WORKSHEET 2

1. sodium hydroxide _____
2. $Zn(NO_2)_2$ _____
3. ammonium chloride _____
4. K_2O_2 _____
5. magnesium bromide _____
6. $Li_2Cr_2O_7$ _____
7. barium acetate _____
8. Cu_3PO_4 _____
9. silver carbonate _____
10. $FeSO_4$ _____
11. nickel (II) nitrate _____
12. $Cr(MnO_4)_3$ _____
13. iron (III) chlorate _____
14. HgO _____
15. aluminum sulfite _____
16. PbI_2 _____
17. copper (II) chromate _____
18. $Hg_2(HSO_4)_2$ _____
19. calcium fluoride _____
20. $Sr(HCO_3)_2$ _____
21. cobalt (II) sulfide _____
22. Li_2SO_3 _____
23. barium iodide _____
24. $CuCrO_4$ _____
25. mercury (I) nitrite _____
26. $CuCl$ _____
27. strontium hydroxide _____
28. Ag_2S _____
29. iron (II) bromide _____
30. $NiCr_2O_7$ _____
31. chromium (III) acetate _____
32. $FePO_4$ _____
33. sodium peroxide _____
34. $HgCO_3$ _____
35. aluminum sulfate _____
36. $Pb(NO_3)_2$ _____
37. ammonium chlorate _____
38. $KMnO_4$ _____
39. magnesium oxide _____
40. $Ca(HSO_4)_2$ _____
41. zinc hydrogen carbonate _____
42. CoF_2 _____
43. nickel (II) nitrite _____
44. $Cr(MnO_4)_3$ _____
45. iron (III) phosphate _____
46. Na_2O_2 _____
47. mercury (II) carbonate _____
48. Al_2S_3 _____
49. ammonium sulfite _____
50. KCl _____
51. magnesium hydroxide _____
52. CaI_2 _____
53. silver chromate _____
54. $Co(NO_3)_2$ _____
55. zinc chlorate _____
56. $CuBr$ _____
57. strontium sulfate _____
58. PbO _____
59. iron (II) hydrogen sulfate _____
60. $LiHCO_3$ _____